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_____ 1. According to
the kinetic-molecular

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theory, particles of an ideal gas a. attract each other but do not collide. b. repel each other and collide.

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Example 1. How many molecules of SO_3 are needed to react with 144 molecules of Fe_2O_3 given this balanced chemical equation?. Fe

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$\text{Fe}_2(\text{SO}_4)_3$. Solution.

We use the balanced chemical equation to construct a conversion factor between Fe_2O_3 and SO_3 . The number of molecules of Fe_2O_3 goes on the bottom of our conversion factor so it cancels with our given amount ...

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reaction in which an
element replaces one
element in a compound
double replacement
reactions the negative
ions in two compounds
switch places, forming
two new compounds

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the amount of products

that form when a

reaction is carried out

in the laboratory.

actual yield, the

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calculation of quantities in chemical equation.

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reactant and product in a chemical reaction would allow you to determine a. the energy released in the reaction. b. the speed of the reaction.

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